

# Developing A Nanotechnology To Support Green Ammonia Production

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Te Herenga Waka—Victoria University of Wellington

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WEDNESDAY SEPTEMBER 6, 6:00PM – 7:00PM

# THE FIVE CLIMATE GRAND CHALLENGES

FROM 51 BILLION TONS PER YEAR TO ZERO



MANUFACTURING: 30%

## How We Make Things



AGRICULTURE: 21%

## How We Grow Things



TRANSPORTATION: 16%

## How We Get Around



BUILDINGS: 7%

## How We Live



ELECTRICITY: 26%

## How We Plug In

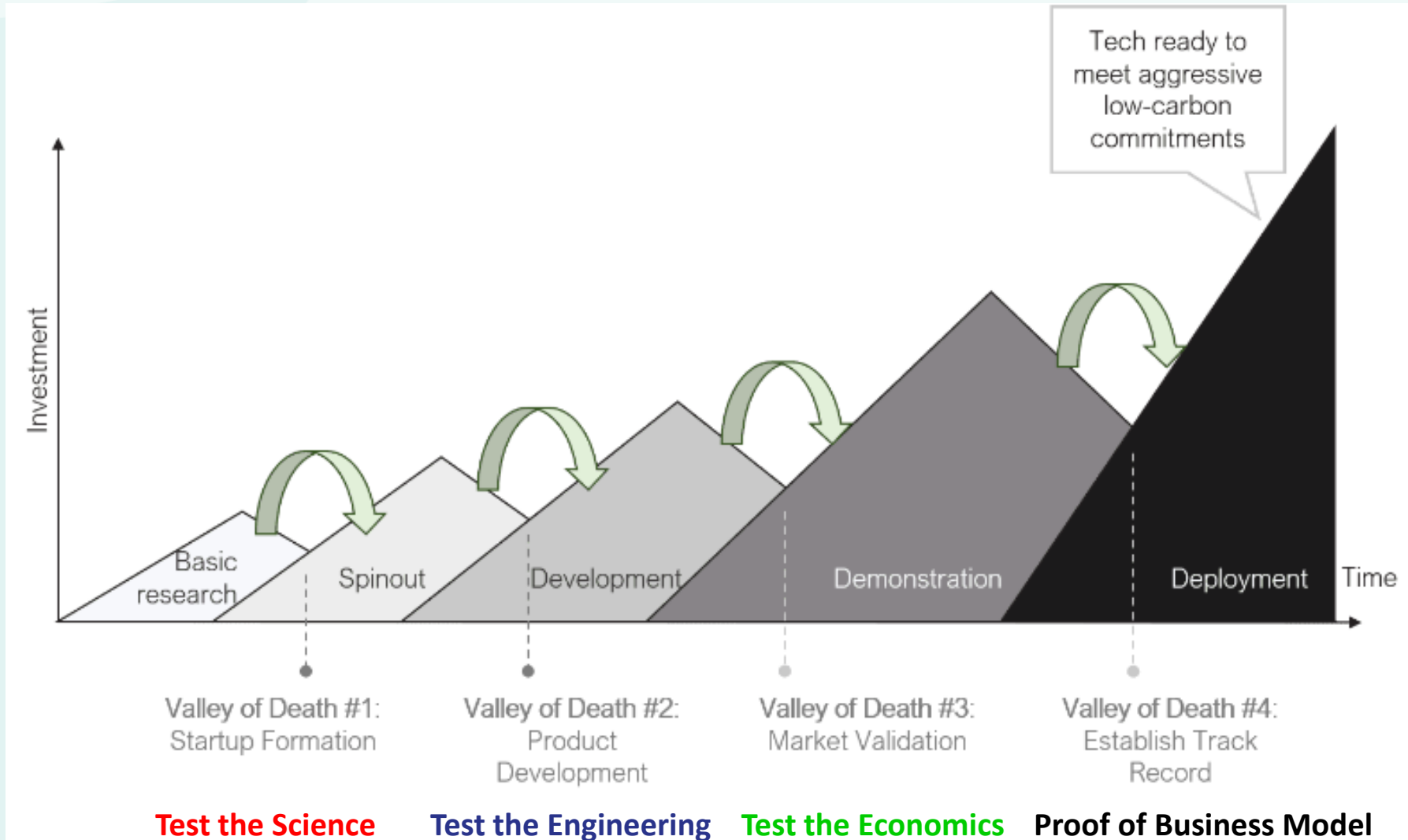
# HEAVY INDUSTRY MANUFACTURING

FROM 51 BILLION TONS PER YEAR TO ZERO



# HEAVY INDUSTRY MANUFACTURING

## LIFECYCLE OF CLIMATE-TECHNOLOGY





# HEAVY INDUSTRY MANUFACTURING

FROM 51 BILLION TONS PER YEAR TO ZERO



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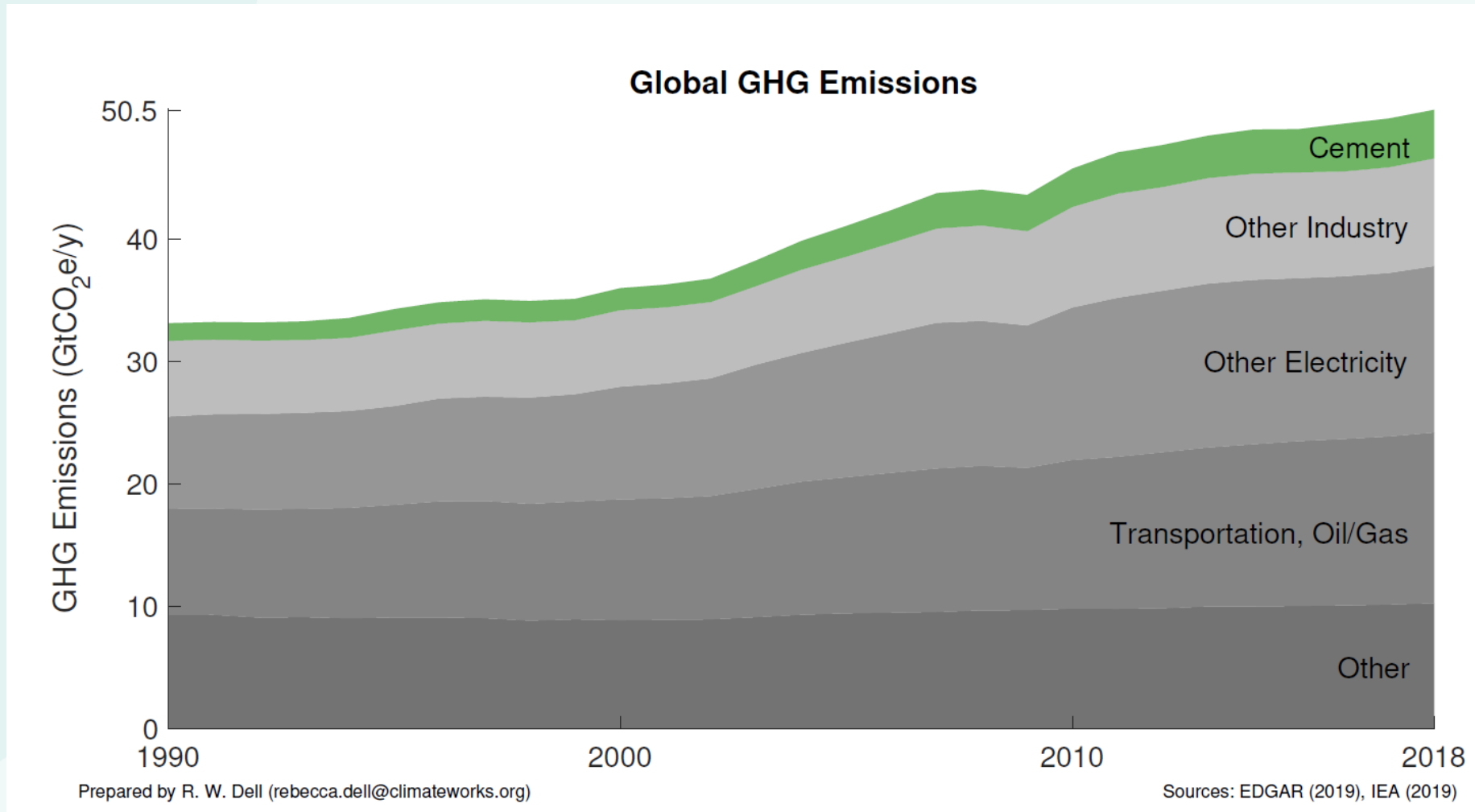
## FIRST EXAMPLE: CEMENT



# HEAVY INDUSTRY MANUFACTURING

## CEMENT

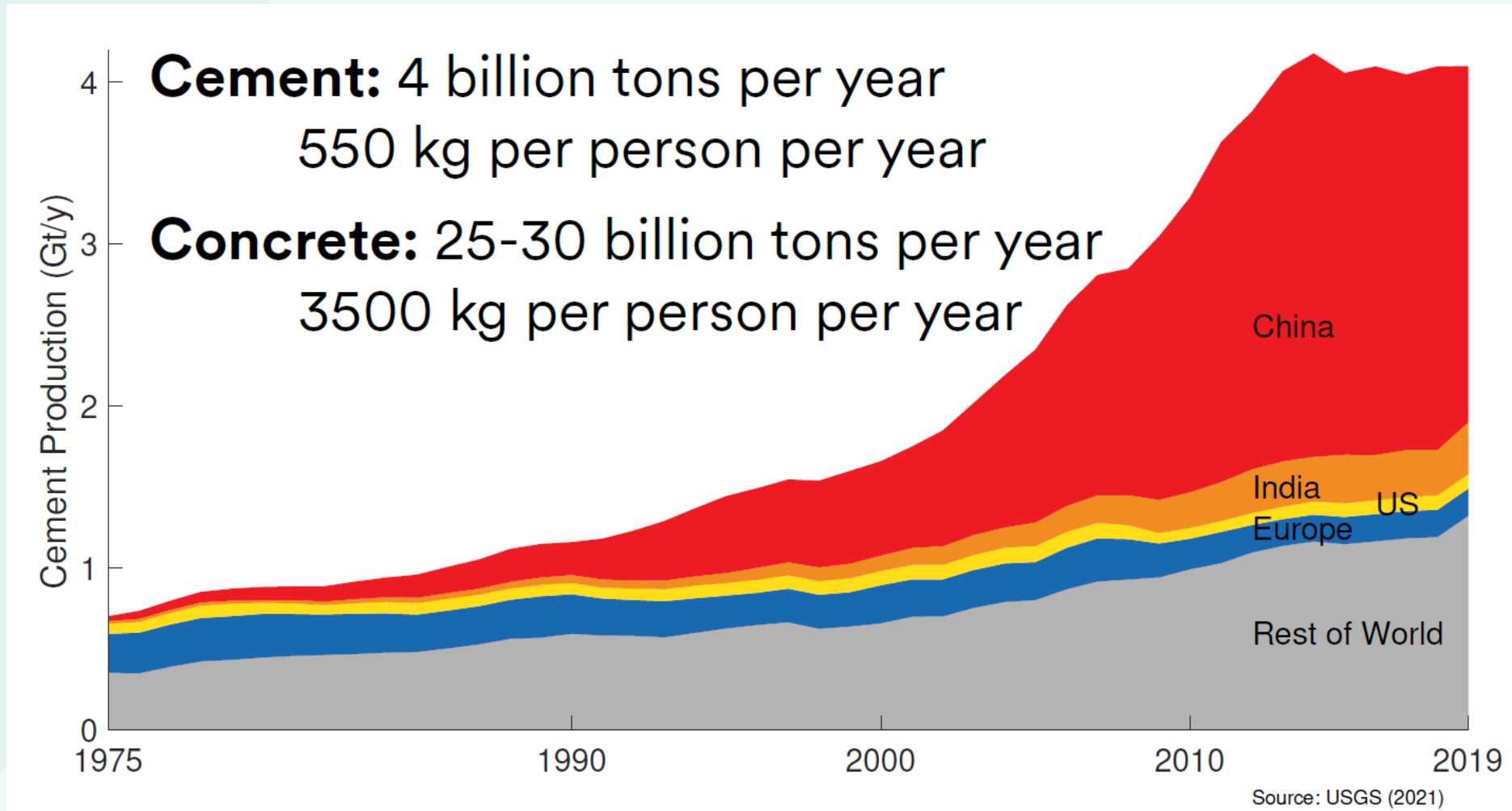
Cement is responsible for about 7% of greenhouse gas emissions.



# HEAVY INDUSTRY MANUFACTURING

## CEMENT

That's because we use an astonishing amount of it!

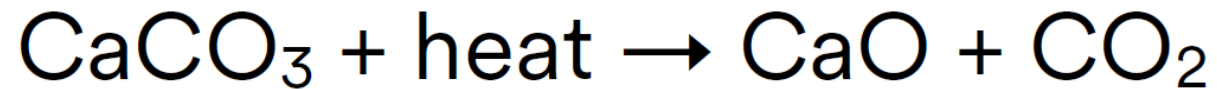
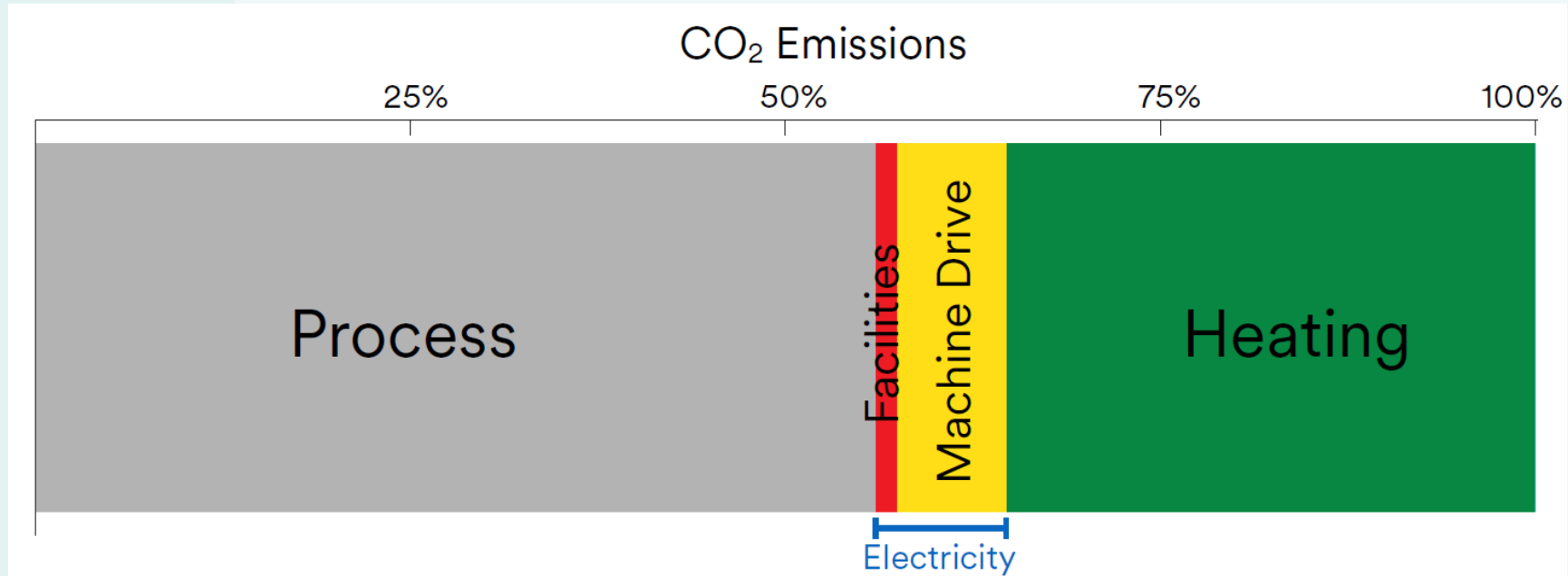




# HEAVY INDUSTRY MANUFACTURING

## CEMENT

Producing cement emits nearly 1 ton CO<sub>2</sub> for each ton of cement!



CaCO<sub>3</sub> = calcium carbonate (Limestone)

CaO = calcium oxide (Cement)

# HEAVY INDUSTRY MANUFACTURING

## CEMENT

Cement is extremely cheap!



**\$1,000**

# HEAVY INDUSTRY MANUFACTURING

FROM 51 BILLION TONS PER YEAR TO ZERO



**SECOND EXAMPLE: AMMONIA**



# HEAVY INDUSTRY MANUFACTURING

AMMONIA=FERTILISERS



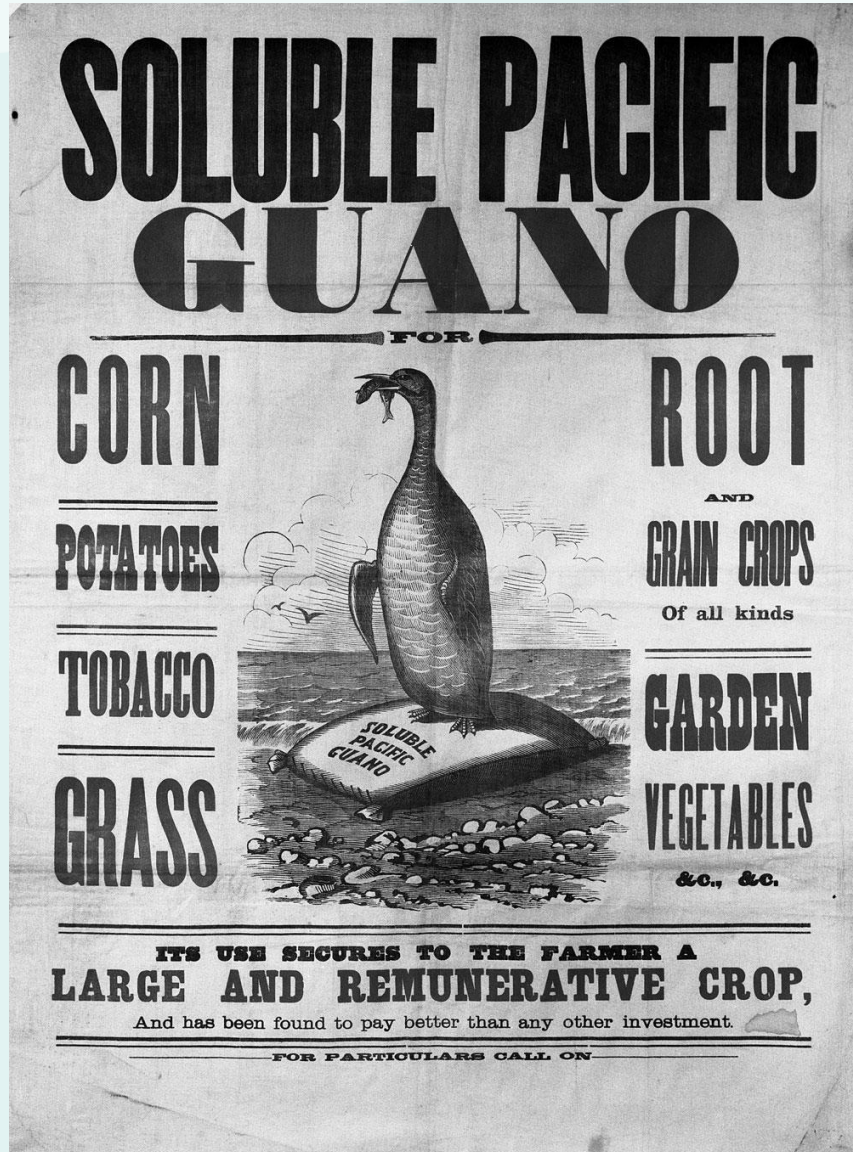
Tractor applying conventional anhydrous ammonia fertilizer on a field (US)



# HEAVY INDUSTRY MANUFACTURING

BEFORE AMMONIA-BASED FERTILISERS: GUANO

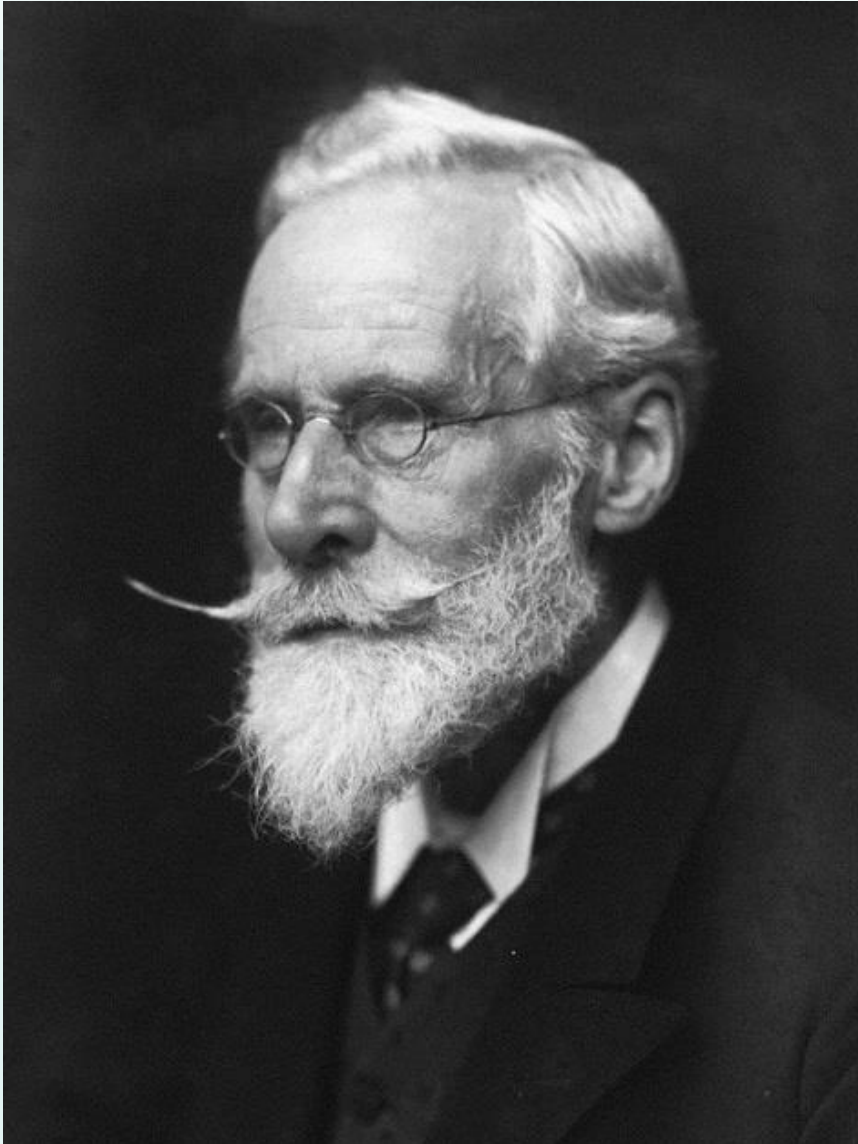
More Precious than Gold: The Story of the Peruvian Guano Trade  
by David Hollett



~1840-1880: Seabird excrement imported from Chincha Islands

# HEAVY INDUSTRY MANUFACTURING

BEFORE AMMONIA-BASED FERTILISERS: LIMITED NATURAL RESSOURCES



*"My chief subject is of interest to the whole world - to every race, to every human being. It is of urgent importance to-day, and it is a life- and - death question for generations to come. I mean the question of food supply. Many of my statements you may think are of the alarmist order; certainly they are depressing, but they are founded on stubborn facts. They show that England and all civilized nations stand in deadly peril of not having enough to eat. As mouths multiply, food resources dwindle. Land is a limited quantity, and the land that will grow wheat is absolutely dependent on difficult and capricious natural phenomena"*

**Sir William Crookes, Bristol, 1898.**

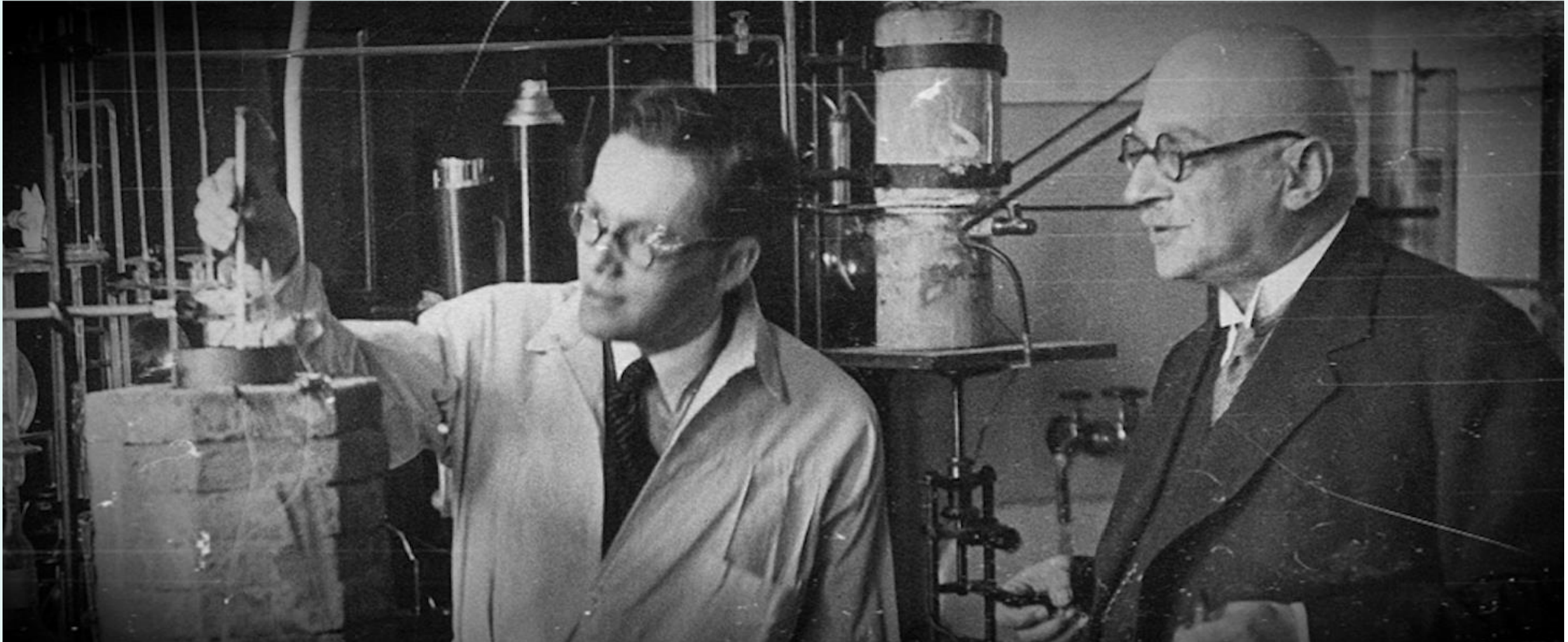
**President of the British Academy of Sciences**



# HEAVY INDUSTRY MANUFACTURING

AMMONIA: SYNTHETIC FERTILISERS

Haber-Bosch: ammonia production process (1920-1930)



# HEAVY INDUSTRY MANUFACTURING

## AMMONIA: SYNTHETIC FERTILISERS

### Haber-Bosch process

“Of all the century's technological marvels, the Haber-Bosch process has made the most difference to our survival” Vaclav Smil

~1920 – Tea spoon of ammonia per day



~2000 – >> 5000 tonnes of ammonia per day



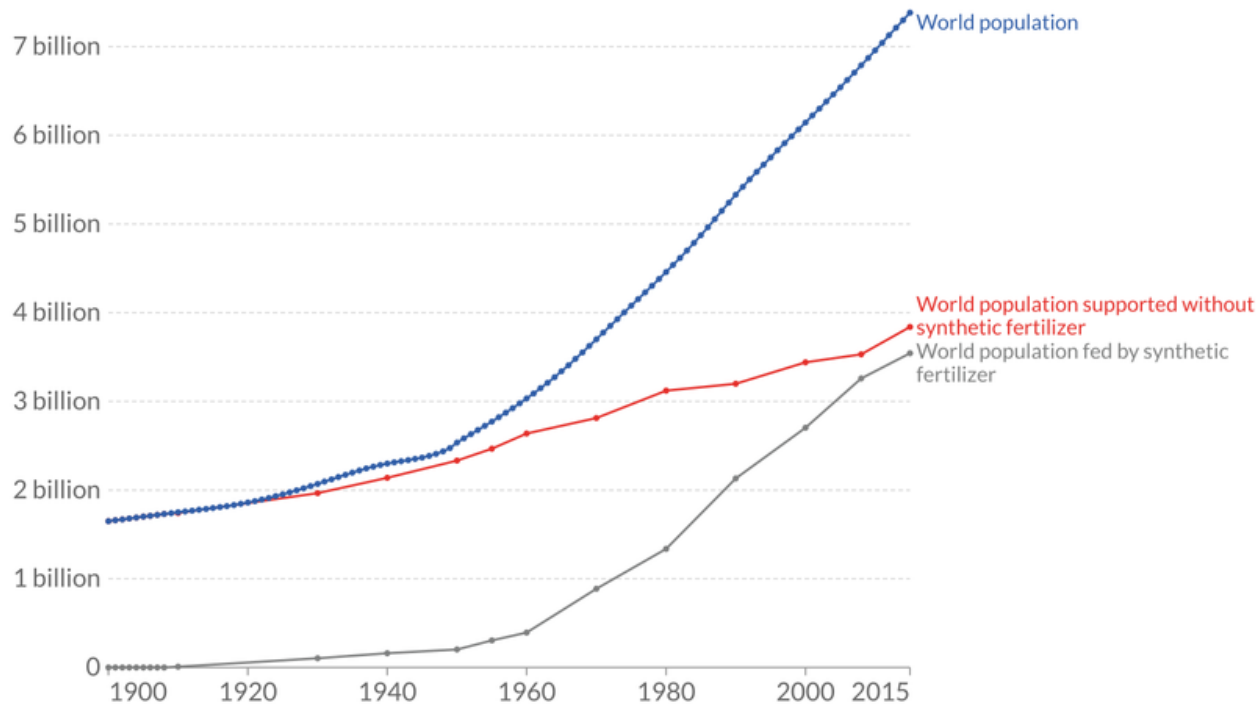


# HEAVY INDUSTRY MANUFACTURING

## AMMONIA: SYNTHETIC FERTILISERS

### World population with and without synthetic nitrogen fertilizers

Estimates of the global population reliant on synthetic nitrogenous fertilizers, produced via the Haber-Bosch process for food production. Best estimates project that just over half of the global population could be sustained without reactive nitrogen fertilizer derived from the Haber-Bosch process.



Source: Erismann et al. (2008); Smil (2002); Stewart (2005)

### Major types of nitrogen-based fertiliser with % nitrogen content:

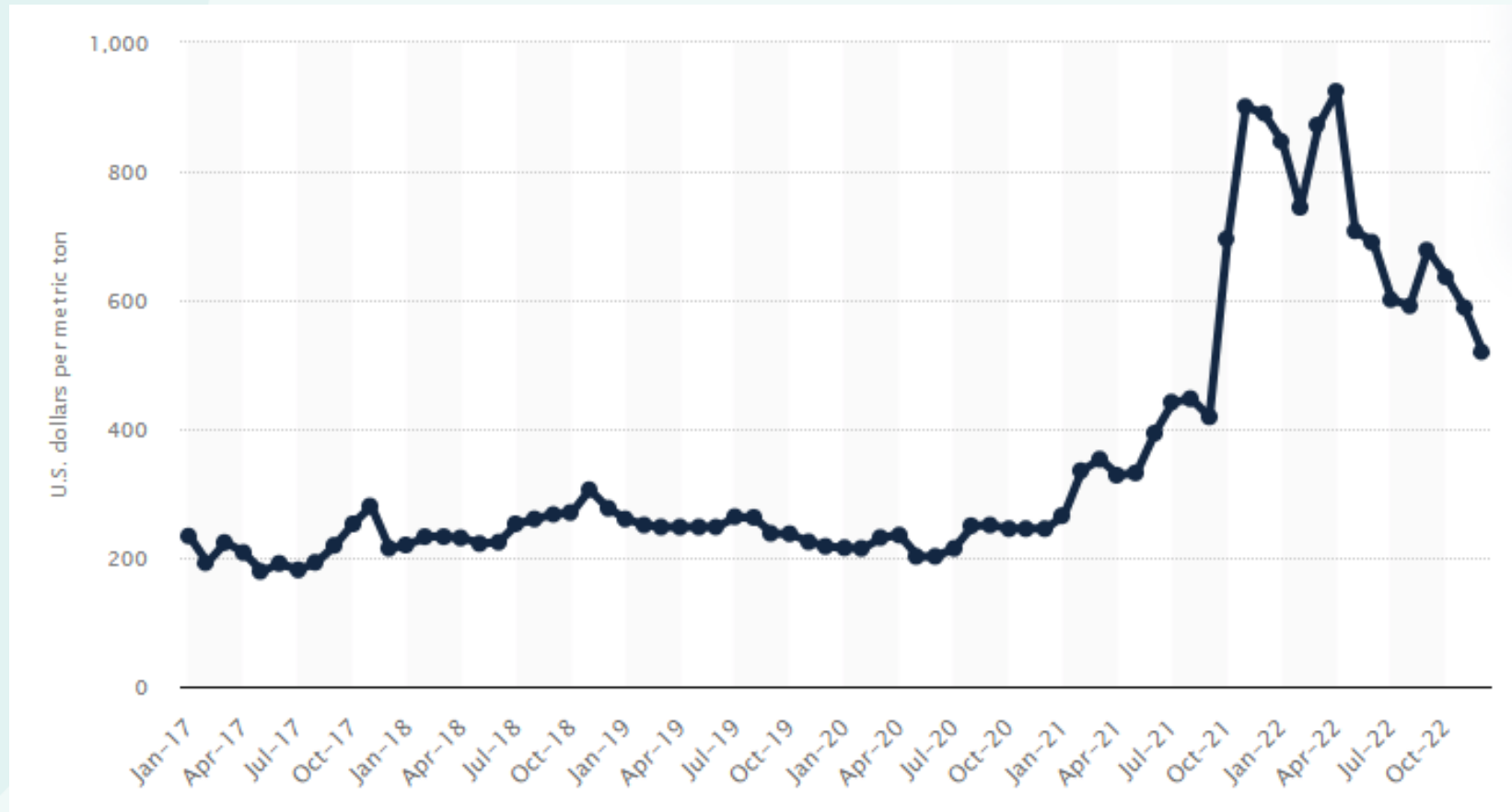
- Urea (46%), anhydrous ammonia (82%), ammonium nitrate (34%), ammonium sulfate (21%), nitrogen solutions (30%), diammonium phosphate (18%).
- Choice is function of cost and equipment available.
- Most common globally is urea.
- Manure (~1% N). Poultry manure has highest N content

If crop yields would have stayed at 1900 levels, we'd need 10x more land for agriculture than we use today

# HEAVY INDUSTRY MANUFACTURING

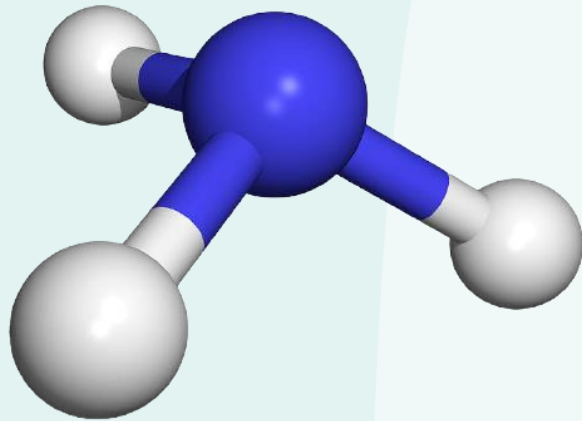
## AMMONIA: SYNTHETIC FERTILISERS

Monthly prices of Urea fertilizer worldwide from January 2017 to December 2022

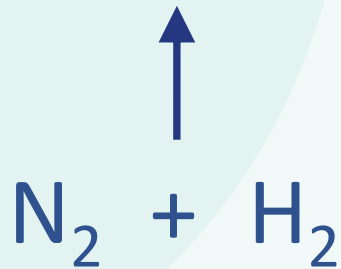


# HEAVY INDUSTRY MANUFACTURING

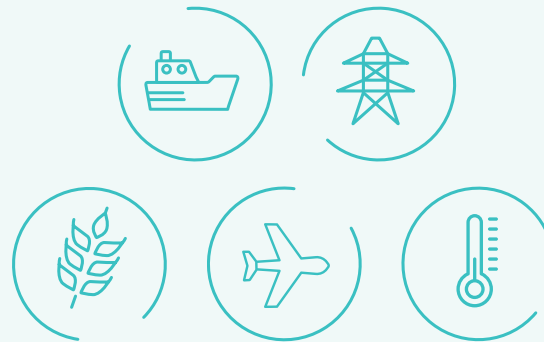
## AMMONIA PRODUCTION



Ammonia:  $\text{NH}_3$



$\text{NH}_3$  is one of largest chemical processes (Haber-Bosch) on the planet



For every tonne of  $\text{NH}_3$  produced, three tonnes of  $\text{CO}_2$  are emitted

~ 1 billion tons of  $\text{CO}_2$  per year

# HEAVY INDUSTRY MANUFACTURING

## THE GREEN AMMONIA CHALLENGE

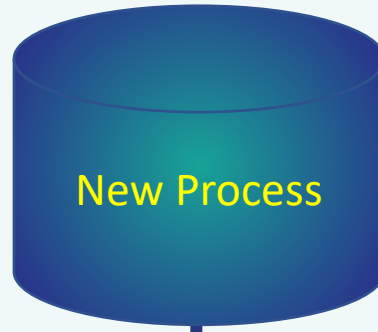
Nitrogen From Air



+



NEED Lower temperature  
Lower pressure synthesis



Catalyst developments

**Breaking the N-N bond is very difficult**

Haber-Bosch Process Needs:

- Iron or ruthenium catalyst
- $T = 400-450\text{ }^{\circ}\text{C}$
- $P = 200-300\text{ atm}$

**Much interest in developing  
new catalysts to avoid these  
harsh conditions**

Hydrogen from Water  
(Renewable energy electrolysis)



# AMMONIA PROCESS AT THE ATOMIC SCALE

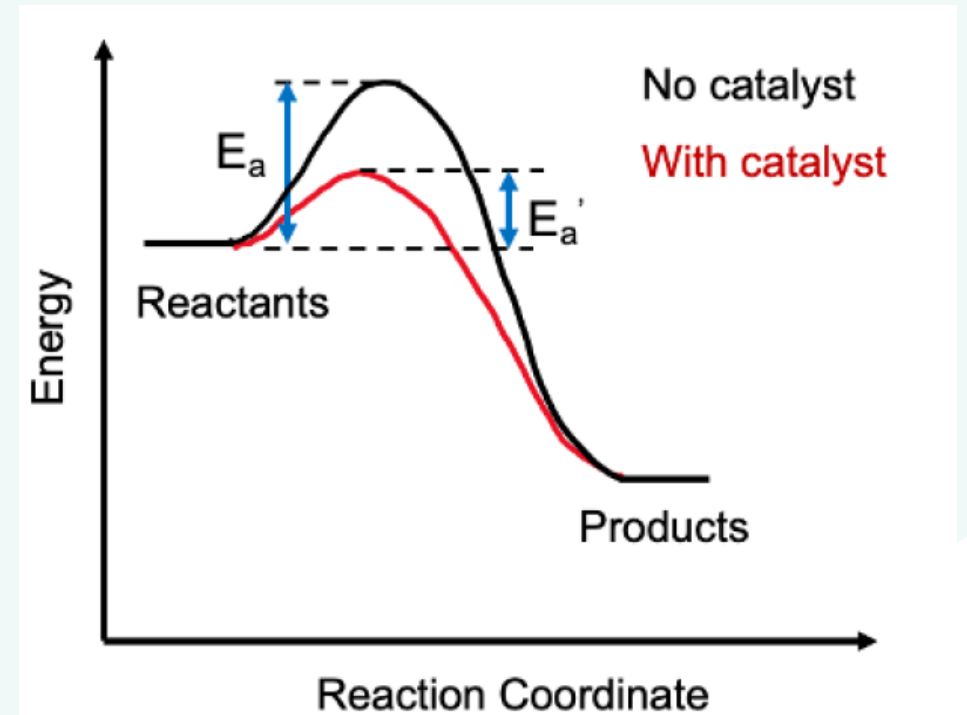
## THE CATALYST: THE CHALLENGE

Iron catalysts



Role Catalysts

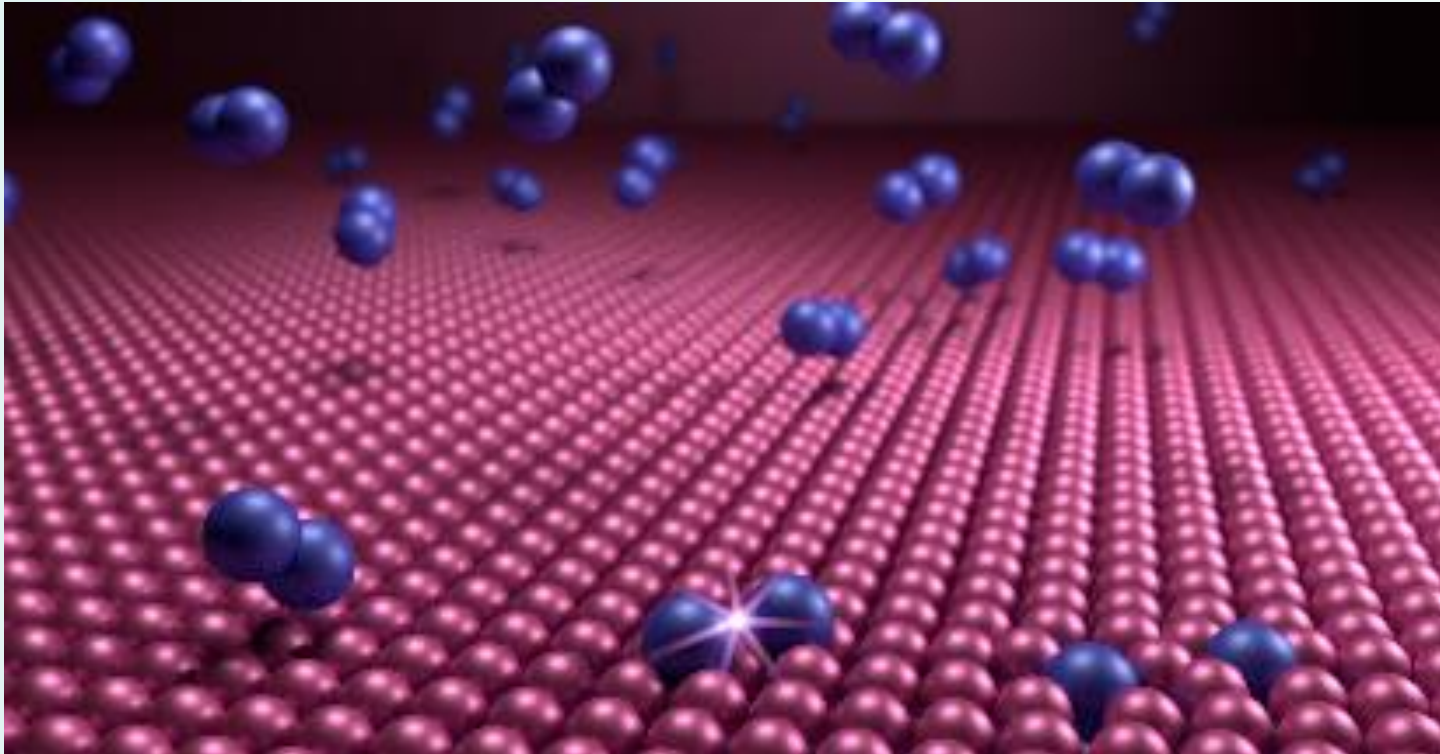
Alternative pathway for the reaction to follow that has that lower activation energy; easier and faster for the reactants ( $\text{H}_2$  &  $\text{N}_2$ ) to form the product ( $\text{NH}_3$ ).



# AMMONIA PROCESS AT THE ATOMIC SCALE

## THE CATALYST: THE CHALLENGE

The Scientific Journey: Find (= fabricate) a material (=catalyst) that can break (=dissociate) molecular nitrogen at mild temperature and pressure



## THE CATALYST: THE CHALLENGE

# Crystal Structures of Elements at STP

STP - Standard Temperature and Pressure

<b>H</b> HEX																	<b>He</b> HCP				
<b>Li</b> BCC	<b>Be</b> HCP	<b>BCC</b> - Body-centered Cubic <b>FCC</b> - Face-centered Cubic <b>HEX</b> - Simple Hexagonal <b>HCP</b> - Close-packed Hexagonal <b>DHCP</b> - Double Close-packed Hexagonal <b>RHO</b> - Rhombohedral										<b>BCT</b> - Body-centered Tetragonal <b>ORTH</b> - Orthorhombic <b>DC</b> - Diamond Cubic <b>DT</b> - Diamond Tetragonal <b>SC</b> - Simple Cubic * predicted crystal structure				<b>B</b> RHO	<b>C</b> HEX	<b>N</b> complex HCP	<b>O</b> P-cubic	<b>F</b> P-cubic	<b>Ne</b> FCC
<b>Na</b> BCC	<b>Mg</b> HCP	<b>K</b> BCC	<b>Ca</b> FCC	<b>Sc</b> HCP	<b>Ti</b> HCP	<b>V</b> BCC	<b>Cr</b> BCC	<b>Mn</b> α-Mn	<b>Fe</b> BCC	<b>Co</b> HCP	<b>Ni</b> FCC	<b>Cu</b> FCC	<b>Zn</b> HCP	<b>Ga</b> complex F-ORTH	<b>Ge</b> DC	<b>As</b> P-RHO	<b>Se</b> complex HEX	<b>Br</b> complex C-ORTH	<b>Kr</b> FCC		
<b>Rb</b> BCC	<b>Sr</b> FCC	<b>Y</b> HCP	<b>Zr</b> HCP	<b>Nb</b> BCC	<b>Mo</b> BCC	<b>Tc</b> HCP	<b>Ru</b> HCP	<b>Rh</b> FCC	<b>Pd</b> FCC	<b>Ag</b> FCC	<b>Cd</b> HCP	<b>In</b> BCT	<b>Sn</b> DT	<b>Sb</b> P-RHO	<b>Te</b> complex HEX	<b>I</b> complex C-ORTH	<b>Xe</b> FCC				
<b>Cs</b> BCC	<b>Ba</b> BCC	57-71	<b>Hf</b> HCP	<b>Ta</b> BCC	<b>W</b> BCC	<b>Re</b> HCP	<b>Os</b> HCP	<b>Ir</b> FCC	<b>Pt</b> FCC	<b>Au</b> FCC	<b>Hg</b> RHO	<b>Tl</b> HCP	<b>Pb</b> FCC	<b>Bi</b> RHO	<b>Po</b> SC	<b>At</b> FCC*	<b>Rn</b> FCC*				
<b>Fr</b> BCC*	<b>Ra</b> BCC	89-103	<b>Rf</b> HCP*	<b>Db</b> BCC*	<b>Sg</b> BCC*	<b>Bh</b> HCP*	<b>Hs</b> HCP*	<b>Mt</b> FCC*	<b>Ds</b> BCC*	<b>Rg</b> BCC*	<b>Cn</b> HCP*	<b>Nh</b> HCP*	<b>Fl</b> FCC*	<b>Mc</b> UNKNOWN	<b>Lv</b> UNKNOWN	<b>Ts</b> UNKNOWN	<b>Og</b> FCC*				

Solid state at STP

Liquid state at STP

Gaseous state at STP

<b>La</b> DHCP	<b>Ce</b> DHCP	<b>Pr</b> DHCP	<b>Nd</b> DHCP	<b>Pm</b> DHCP	<b>Sm</b> complex RHO	<b>Eu</b> BCC	<b>Gd</b> HCP	<b>Tb</b> HCP	<b>Dy</b> HCP	<b>Ho</b> HCP	<b>Er</b> HCP	<b>Tm</b> HCP	<b>Yb</b> FCC	<b>Lu</b> HCP
<b>Ac</b> FCC	<b>Th</b> FCC	<b>Pa</b> BCT	<b>U</b> ORTH	<b>Np</b> ORTH	<b>Pu</b> MONO	<b>Am</b> DHCP	<b>Cm</b> DHCP	<b>Bk</b> DHCP	<b>Cf</b> DHCP	<b>Es</b> FCC	<b>Fm</b> FCC*	<b>Md</b> FCC*	<b>No</b> FCC*	<b>Lr</b> HCP*

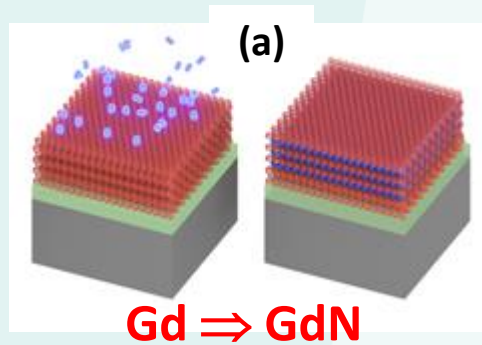
## Lanthanide Materials

# AMMONIA PROCESS AT THE ATOMIC SCALE

## FACILE DISSOCIATION OF $N_2$ BY LANTHANIDE SURFACES

In-situ and real time monitoring of the nitridation of lanthanides - exptl results

- Steps by steps: (2) Exposure to  $N_2$  AND (3) Formation of Gadolinium Nitride surface layer



- (a) Exposure of the Gd surface to  $N_2$ ; at ambient temperature and  $N_2$  partial pressure of  $3 \times 10^{-5}$  mBar.

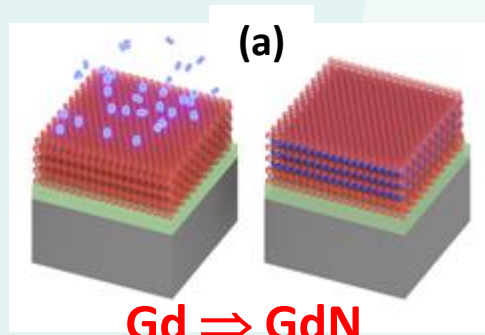


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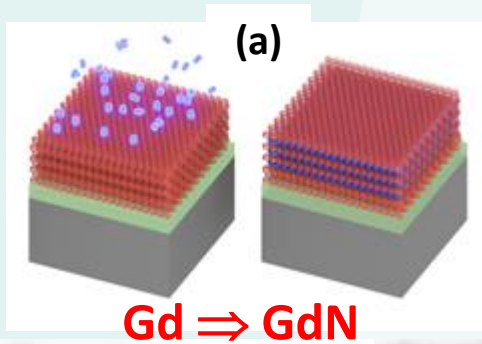
- (a) Exposure of the Gd surface to  $N_2$ ; at ambient temperature and  $N_2$  partial pressure of  $3 \times 10^{-5}$  mBar.
- (b) and (c) RHEED patterns of Gd before (a) and after (c) exposure to  $N_2$ .  
Streak spacing increases, indicating a contraction of the surface Gd lattice spacing;  $a_{GdN} = 3.53 \text{ \AA}$ .

# AMMONIA PROCESS AT THE ATOMIC SCALE

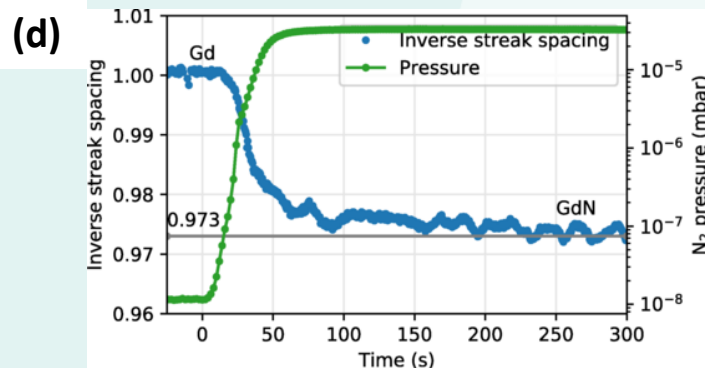
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Streak spacing increases, indicating a contraction of the surface Gd lattice spacing;  $a_{GdN} = 3.53 \text{ \AA}$ .
- (c) full nitridation of the surface within 300 seconds; relatively fast process



# AMMONIA PROCESS AT THE ATOMIC SCALE

TUAKANA-TEINA

SUPPORTING & GROWING PHD STUDENTS & SUPERVISORS



“ It’s been really interesting to work on a project that has real-world application, and to see first-hand how commercialising research can get it out of the lab and into the hands of others where it can make a difference. ”

**JAY CHAN, PHD STUDENT (2014-2018) – Funded Royal Society of New Zealand (Marsden Grant)**  
**Electron diffraction Software Licensed to US-based Company (>X00,00NZD - 2021)**








# AMMONIA PROCESS AT THE ATOMIC SCALE

## TOWARDS AMBIENT TEMPERATURE AMMONIA SYNTHESIS

### Ammonia Synthesis – exptl results

PHYSICAL REVIEW MATERIALS **4**, 115003 (2020)

#### **Facile dissociation of molecular nitrogen using lanthanide surfaces: Towards ambient temperature ammonia synthesis**

J. R. Chan <sup>1,\*</sup> S. G. Lambie <sup>2</sup> H. J. Trodahl <sup>1</sup> D. Lefebvre,<sup>1</sup> M. Le Ster,<sup>3</sup> A. Shaib <sup>1</sup> F. Ullstad,<sup>1</sup> S. A. Brown <sup>3</sup>  
B. J. Ruck,<sup>1</sup> A. L. Garden <sup>2</sup> and F. Natali <sup>1,†</sup>

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Victoria University of Wellington, PO Box 600, Wellington, New Zealand*

<sup>2</sup>*The MacDiarmid Institute for Advanced Materials and Nanotechnology, Department of Chemistry, University of Otago, P.O. Box 56,  
Dunedin 9054, New Zealand*

<sup>3</sup>*The MacDiarmid Institute for Advanced Materials and Nanotechnology, School of Physical and Chemical Sciences,  
University of Canterbury, Private Bag 4800, Christchurch 8140, New Zealand*



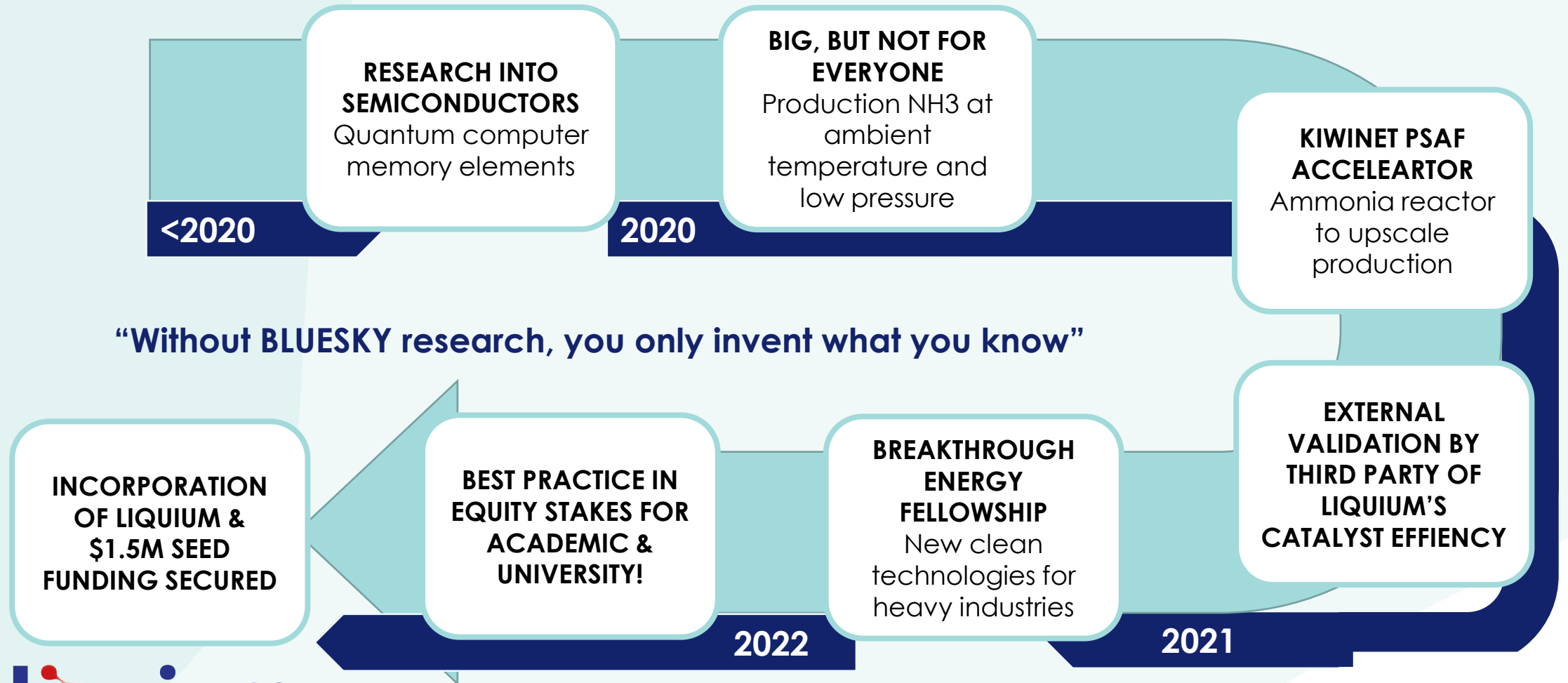
(Received 7 May 2020; revised 22 September 2020; accepted 29 October 2020; published 30 November 2020)

A combined experimental and computational study is reported on a hitherto unrecognised single lanthanide catalyst for the breaking of molecular nitrogen and formation of ammonia at ambient temperature and low pressure. We combine *in situ* electrical conductance and electron diffraction measurements to track the conversion from the lanthanide metals to the insulating lanthanide nitrides. The efficiency of the conversion is then interpreted using DFT+*U* calculations, suggesting a molecular nitrogen dissociation pathway separate from that well established for transition metals. Finally, we show that exposure of the lanthanide surfaces to both molecular nitrogen and hydrogen results in the formation of ammonia.



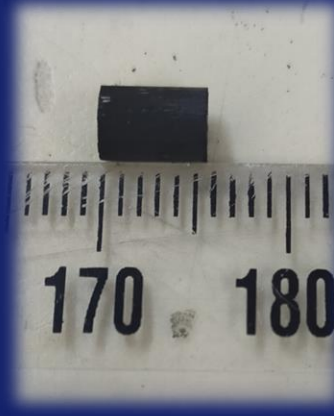
# Bluesky Research To Business Academic

THE JOURNEY



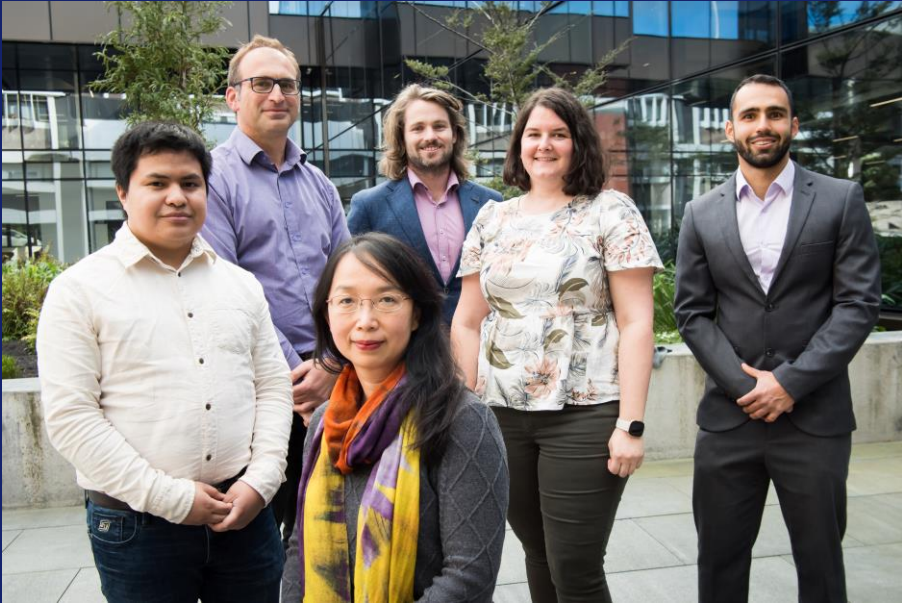
**What makes an industrially  
relevant catalyst?**

# Materials showcase





# Team showcase



- 5 Graduate Students from Victoria University of Wellington,
- 1 Graduate Student from University of Canterbury, Paul Geraghty (Liquium's CEO),
- Pipeline for graduate students,
- Populating the deep tech sector.

# INNOVATION & IMPACT POTENTIAL

## MARKET AND END USERS



# INNOVATION & IMPACT POTENTIAL

## MARKET AND END USERS



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Monday, July 25, 2022

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## World's First Ammonia-fuel Ready Vessel Delivered

MarineLink February 7, 2022



Chinese shipyard New Times Shipbuilding has delivered what is said to be the first ammonia-fuel ready vessel in the world.

The newly built Kiriti Future, recently handed over to owner Avin International, is a 274-meter-long, 156,500 DWT Suezmax tanker classed by ABS and flying the Greek flag.

The ship is currently conventionally fueled but complies with the ABS Ammonia Ready Level 1 requirements, indicating it is designed to be converted to run on ammonia in the future. The vessel also meets the ABS LNG Fuel Ready Level 1 requirements.



# ACKNOWLEDGEMENTS

“SCIENCE IS PEOPLE” - Alan MacDiarmid Nobel Prize (2000, Chemistry)

“A PLACE WHERE TALENT WANTS TO LIVE” - *(Sir Paul Callaghan)*



We **STILL** need to  
go from 51 Billion  
tons of emissions  
per year to Zero